Risk assessment for the benzene leakage from a sunken ship (沈没船からのベンゼン流出のリスク評価)

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# A table of contents

#### **1. Introduction**

1.1. Hazard map of wrecks around

Japan

1.2. An accident outline

1.3. Location

1.4. A field research in Taiwan Strait

**1.5.** The purpose of this research

2. MEC Ocean Model

3. Benzene Model

**3.1.** Governing equation

**3.2. Model verification** 

- 3.3. Computational and boundary conditions
- 3.4. Simulations of benzene spill behavior

4. Risk assessment

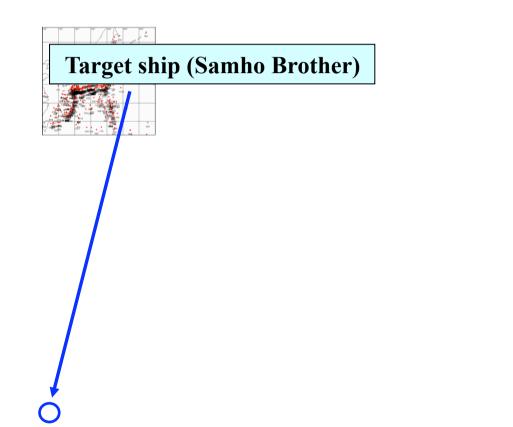
4.1. Introduction

- 4.2. Methods
- 4.3. Results of numerical simulation

**5.** Conclusions



### 1.1. Hazard map of wrecks around Japan



There are about 1,200 wrecks (over 100 tons) around Japan . In USA, 150,000 wrecks are reported.

Fig. 1. Wreck data around Northeast Asia (Kuroda 2006) Notes: 1. Red circle: Position , 2. Black number: Hazard rank

### **1.2.** An accident outline

1. 10<sup>th</sup> Oct, 2005

Samho Brother (Chemical Tanker: 3,100 tons of Benzene, UN1114) capsized off the coast of Taoyuan after collision with another ship.

2. 27th Oct, 2005

Military actions to burn off the benzene by firing missiles at the ship → Failed → Benzene spilled from the vessel at unknown rate

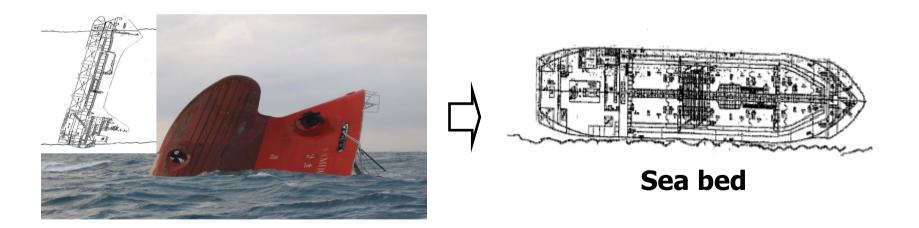


Fig. 2. Before and after military actions

1.3. Location

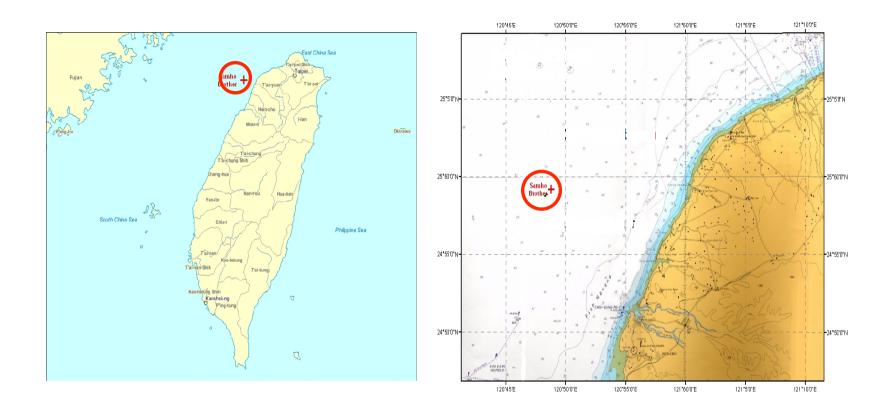


Fig. 3. An accident position: 24.57N-120.48E

(70 kilometers offshore and 75 meters depth)

## **1.4. A field research in Taiwan Strait**



 On 26<sup>th</sup> April, 2007, research teams from Japan and Taiwan launched to measure benzene → Failed or not leaking ?

♪



Fig. 4. (a) Injured fish (b) Installing ADCP (c) Water sampler

## **1.5.** The purpose of this research

- Obtaining the useful data for the planning of the optimal observation
- To support fundamental data for Risk Evaluation and Risk Management



Forecasting model for benzene spills transport and fate

# **2. MEC Ocean Model**

#### • Continuity equation

$$\frac{\partial u}{\partial x} + \frac{\partial v}{\partial y} + \frac{\partial w}{\partial z} = 0$$
(1)

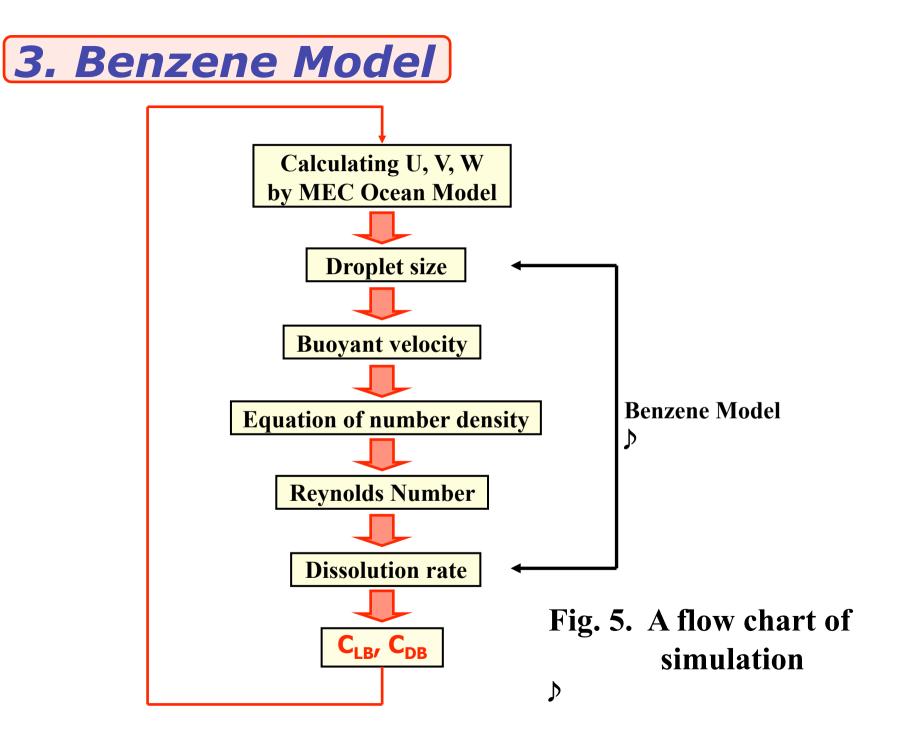
#### • Momentum equations in the x and y directions

$$\frac{\partial u}{\partial t} + u \frac{\partial u}{\partial x} + v \frac{\partial u}{\partial y} + w \frac{\partial u}{\partial z} = -\frac{1}{\rho} \frac{\partial p_s}{\partial x} + f v + A_M \left( \frac{\partial^2 u}{\partial x^2} + \frac{\partial^2 u}{\partial y^2} \right) + \frac{\partial}{\partial z} \left( K_M \frac{\partial u}{\partial z} \right)$$
(2)

$$\frac{\partial v}{\partial t} + u \frac{\partial v}{\partial x} + v \frac{\partial v}{\partial y} + w \frac{\partial v}{\partial z} = -\frac{1}{\rho} \frac{\partial p_s}{\partial y} - f u + A_M \left( \frac{\partial^2 v}{\partial x^2} + \frac{\partial^2 v}{\partial y^2} \right) + \frac{\partial}{\partial z} \left( K_M \frac{\partial v}{\partial z} \right)$$
(3)

• Hydrostatic pressure equation

$$\mathbf{0} = -\frac{1}{\rho} \frac{\partial p_s}{\partial z} - g \tag{4}$$



# 3.1. Equations

#### • Dissolution rate

$$S_{mass} = \frac{dV}{dt} = -\frac{1}{\rho_B} \left( A \frac{ShD}{d} (B_s - B_o) \right)$$
(5)

• Buoyant velocity

$$\frac{dZ}{dt} = \omega = \sqrt{\frac{4d(\rho_s - \rho_B)g}{3C_D\rho_s}}$$
(6)

#### • Equation of number density

$$\frac{\partial n}{\partial t} + \frac{\partial (un)}{\partial x} + \frac{\partial (vn)}{\partial y} + \frac{\partial (wn)}{\partial z} = \frac{\partial (un)}{\partial x} + \frac{\partial (un)}{\partial y} + \frac{\partial (un)}{\partial z} + \frac{\partial (un$$

# 3.1. Equations

• Evaporation rate (Mackay, D. et al., 1983)

$$N = K_{OL} \cdot \left(C_{DB} - \frac{p}{H}\right)$$
(8)  

$$\frac{1}{K_{OL}} = \frac{1}{K_L} + \frac{R \cdot T}{H \cdot K_G}$$
(9)  

$$K_G = 1.0 \times 10^{-3} + 46.2 \times 10^{-3} \times U^* \times Sc_G^{-0.67}$$
(10)  

$$K_L = 1.0 \times 10^{-6} + 34.1 \times 10^{-4} \times U^* \times Sc_L^{-0.5} (U^* > 0.3)$$
(10)  

$$= 1.0 \times 10^{-6} + 144 \times 10^{-4} \times (U^*)^{2.2} \times Sc_L^{-0.5} (U^* < 0.3)$$
(11)  

$$U^* = U_{10} (6.1 + 0.63U_{10})^{0.5}, \text{ outside}$$

#### • Benzene transport equation

$$\frac{\partial C_{LB}}{\partial t} + \frac{\partial (uC_{LB})}{\partial x} + \frac{\partial (vC_{LB})}{\partial y} + \frac{\partial (wC_{LB})}{\partial z} = A_M \left( \frac{\partial^2 C_{LB}}{\partial x^2} + \frac{\partial^2 C_{LB}}{\partial y^2} \right) + \frac{\partial}{\partial z} \left( K_M \frac{\partial C_{LB}}{\partial z} \right) - \frac{\partial}{\partial z} \left( K_M \frac{\partial C_{LB}}{\partial z} \right) + \sum_{lB} (10)$$

$$\frac{\partial C_{DB}}{\partial t} + \frac{\partial (uC_{DB})}{\partial x} + \frac{\partial (vC_{DB})}{\partial y} + \frac{\partial (wC_{DB})}{\partial z} = A_M \left( \frac{\partial^2 C_{DB}}{\partial x^2} + \frac{\partial^2 C_{DB}}{\partial y^2} \right) + \frac{\partial}{\partial z} \left( K_M \frac{\partial C_{DB}}{\partial z} \right) + \sum_{lB} (11)$$

where C<sub>LB</sub> and C<sub>DB</sub> are concentrations of the liquid and dissolved benzene, respectively

# 3.2. Model verification

These are the first and only known field experiments for subsurface oil jet/plumes ( Rye et al., 1997).

Method

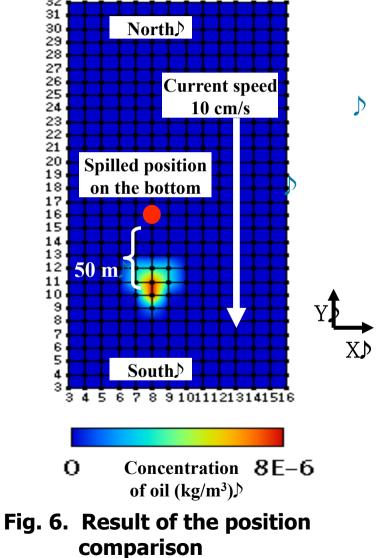
1. Comparing with time that the first

droplet reaches the surface

2. The position comparison of oil jet/p lume at a depth of 3 meters after 1 2 min 30 sec

A result of Method 1.		
Yapa's model	636 sec	
Field experiment	600 sec	
Present model	618 sec	

A result of Method 2.Field experiment40m from an originPresent model50m (Fig. 6.)



#### 3.3. Computational and boundary conditions

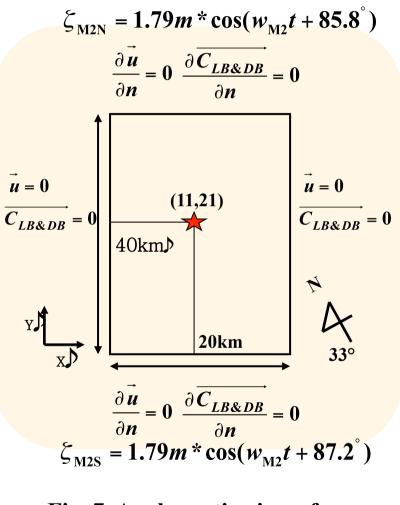


Fig. 7. A schematic view of calculation domain

#### **Parameters for computation**

- 1. Grid size: 1000m
- 2. Time step: 5sec
- 3. Calculation time: 24days
- 4. Beginning of leaking: 3<sup>rd</sup> to 4<sup>th</sup> day
- 5. Number of grids: 20 \* 40 \* 17
- 6. Water depth: 75m (Uniform)
- 7. Calculation area: 20km by 40km

Table 1 Benzene release scenarios

<b>ب</b>	Wind speed (m/s)	Leaking rate (kg/s)	Droplet size (m)	Duration per iod (day)
1	5 (Summer)	10	0.01	1
2	10 (Winter)	10	0.01	1
3	5	10	0.005	1
4	10	10	0.005	1
5	5	1	0.005	1
6	10	1	0.005	1
7	5	1	0.01	1
8	10	1	0.01	1

• Changes of droplet size, buoyant velocity, and dissolution rate with 8 cases (Table 1)

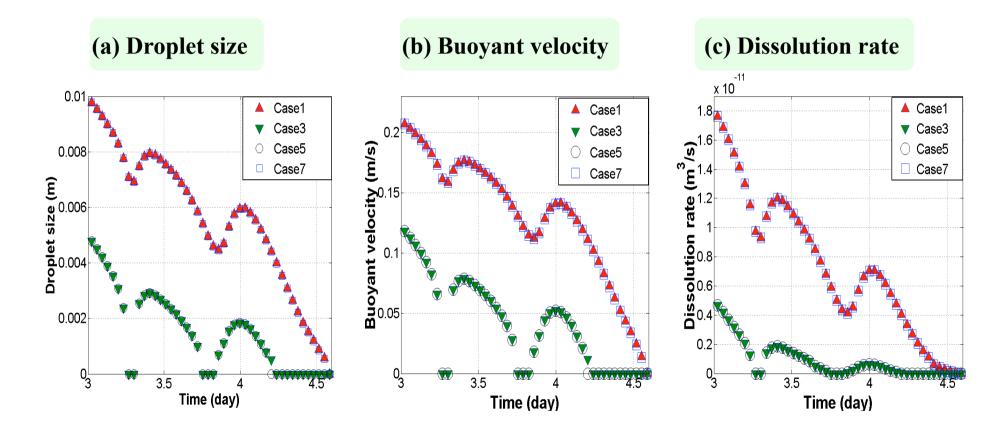


Fig. 8. Time history profile in an accident position at 3 m depth: grid point (11,21)

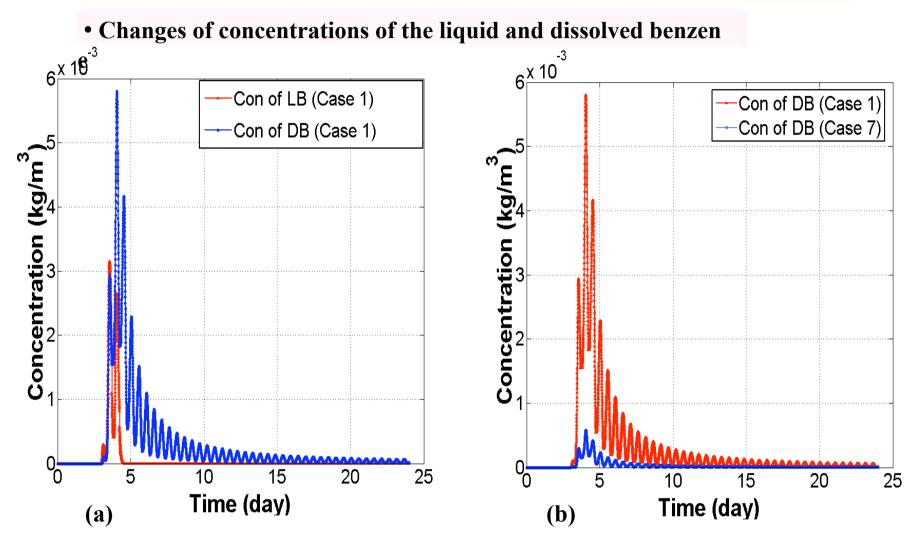


Fig. 9. Time history profiles of concentrations of the liquid and dissolved benzene in Case 1 (a) and concentrations of the dissolved benzene in Case 1(10kg/s) and Case 7(1kg/s) in an accident position at 3 m depth: grid point (11,21)

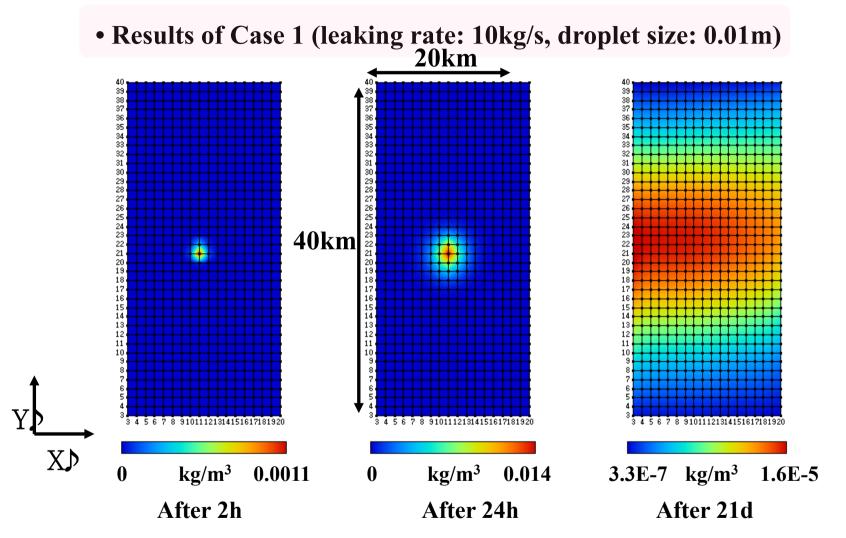


Fig. 10. Horizontal distributions of concentration of the dissolved benzene at 3 m depth
Note: Color bar: Concentration of the dissolved benzene (kg/m<sup>3</sup>)



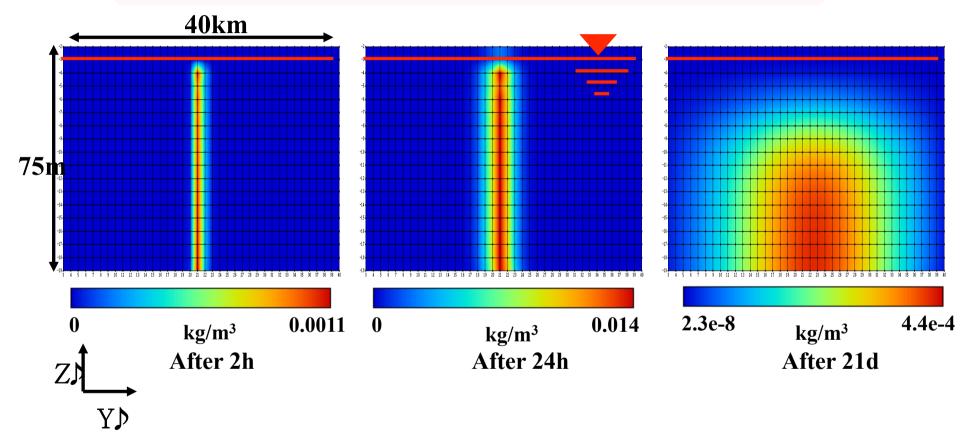


Fig. 11. Cross-sectional views of concentration of the dissolved benzene at 3 m depth Note: Color bar: Concentration of the dissolved benzene (kg/m<sup>3</sup>)

#### • Tidal effects on the subsurface benzene spills

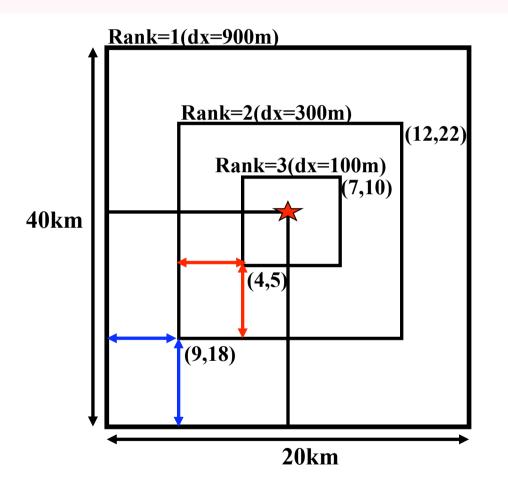


Fig. 12. A nested grid system

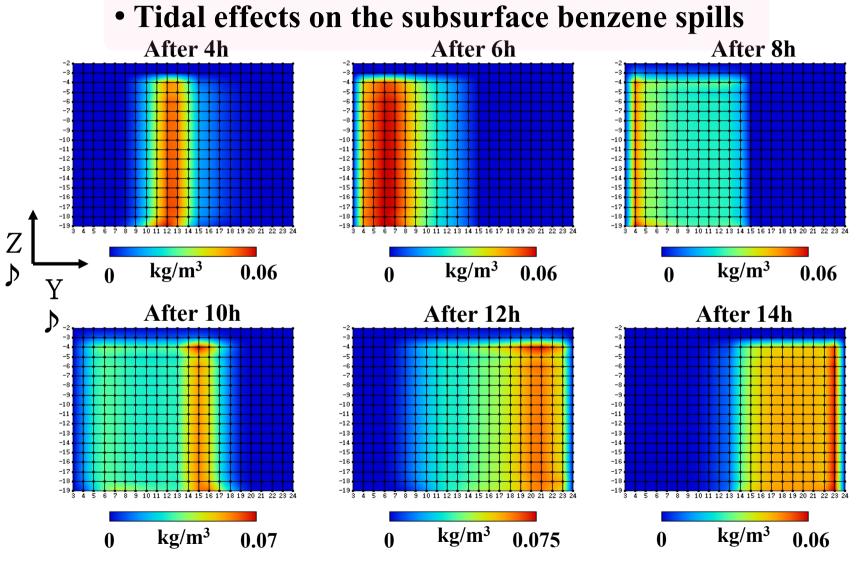
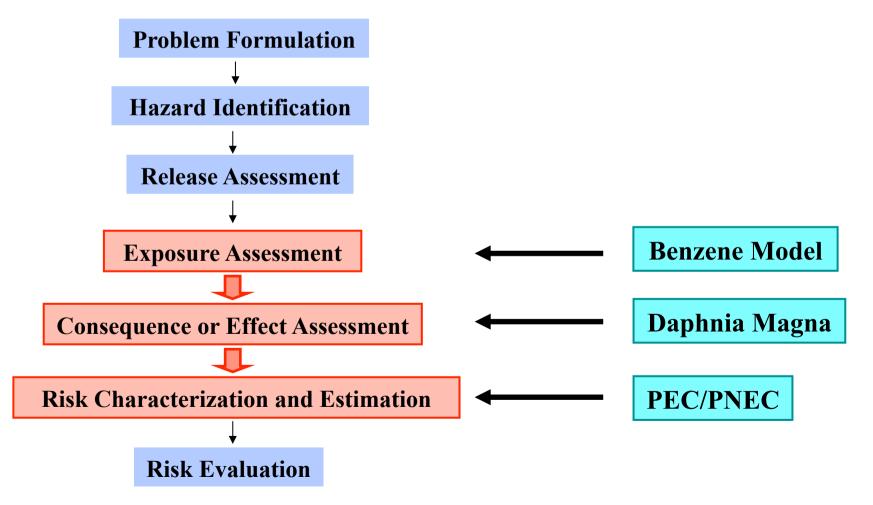
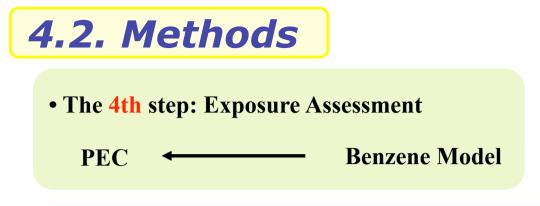


Fig. 13. Cross-sectional views of the concentration of the dissolved benzene



In general seven steps can be identified addressing the key questions in an Ecological Risk Assessment (ERA) (Fairman R. et el., 1999)





#### • The 5th step: Consequence or Effect Assessment

PNEC values with Assessessment Factors(AF) of 100(OECD) or 1000(EU) (Tabel 2)

Daphnia Magna is chosen for ecotoxicity tests.

 Table 2 Defined PNEC values [9]



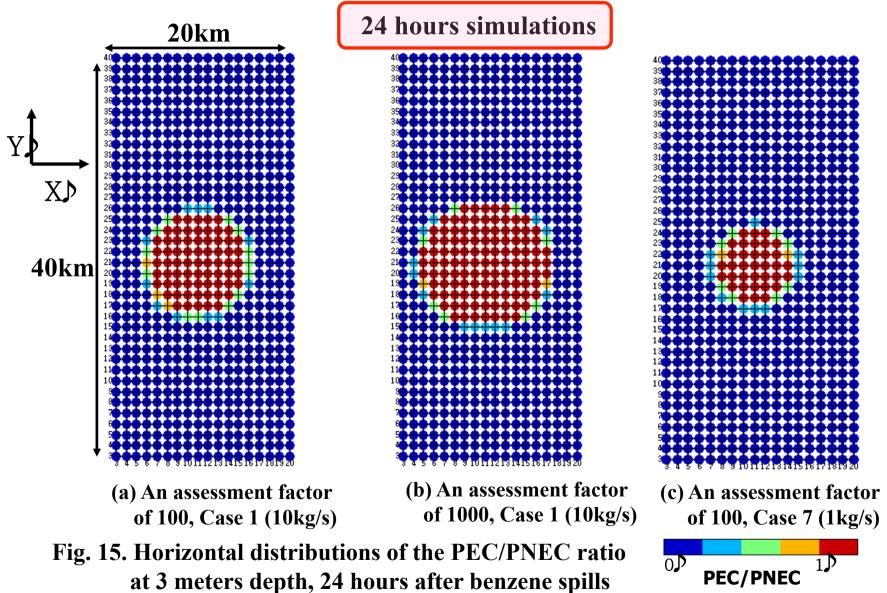
Fig. 14. A photo of Daphnia Magna

	24hr	21d
	(acute)	(chronic)
100	0.08	0.01
(AF)	(mg/l)	(mg/l)
1000	0.008	0.001
(AF)	(mg/l)	(mg/l)

• The 6th step: Risk Characterization and Estimation

PEC (Predicted Exposure Concentration)PNEC (Predicted No Effective Concentration)

# 4.3. Results of numerical simulations



Note: Color bar: PEC/PNEC ratio

# 4.3. Results of numerical simulations

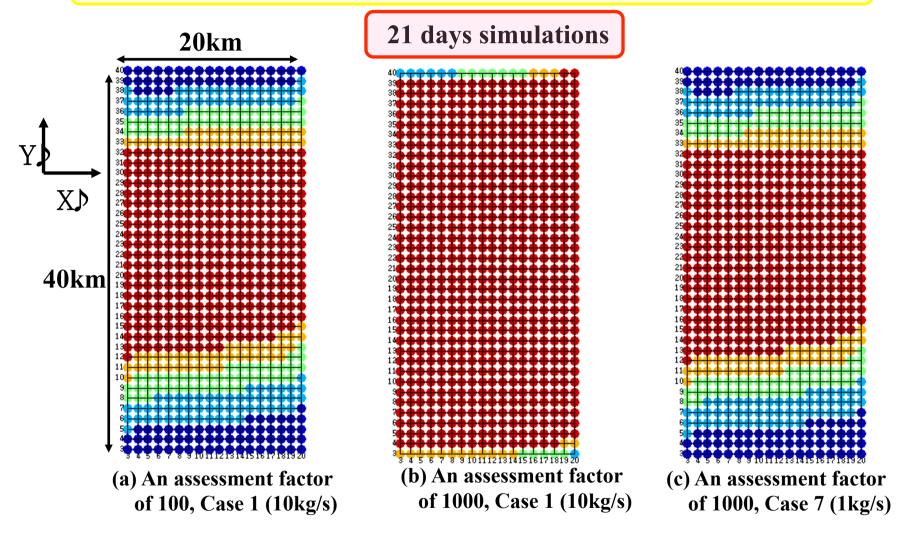
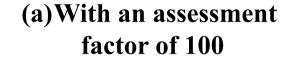


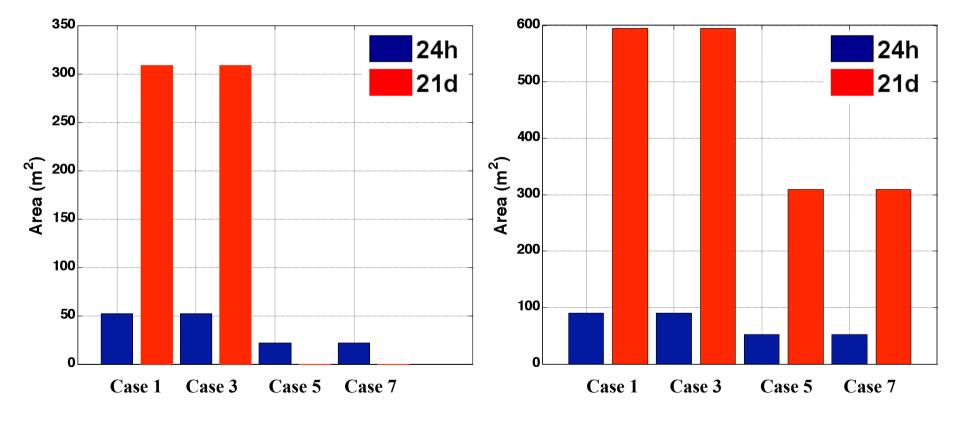
Fig. 16. Horizontal distributions of the PEC/PNEC ratio at 3 meters depth, 21 days after benzene spills Note: Color bar: PEC/PNEC ratio



4.3. Results of numerical simulations



(b) With an assessment fa ctor of 1000



#### Fig. 17. Histogram of polluted areas which PEC/PNEC ratio is over 1 at 3 meters depth

# 5. Conclusions

- In order to simulate benzene diffusion in Taiwan Strait, Benzene Model is developed.
- Comparing with observation data of oil jet/plume, the present model agrees well.
- An important factor to determine buoyant velocity, dissolution rate, and Reynolds number is the initial droplet size.
- Leaking rate and spill duration time strongly affect the concentration of the dissolved benzene in water.
- Through the results of the risk assessment, the casualties would be tremendous.

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# Thank you for your attention!!!